## SYNTHESIS AND PROPERTIES OF UNSATURATED THIETANES

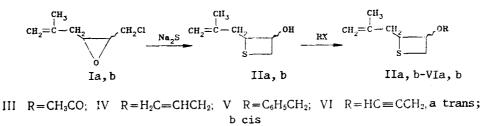
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Reaction of unsaturated epichlorohydrins with sodium sulfide in aqueous-alcoholic solution leads to 3-hydroxythietanes. Ethers of the latter can be prepared by phase transfer catalysis using acetic anhydride and alkyl halides.

Thietanes can be obtained from thioepichlorohydrin and potassium hydroxide [1] but also via aliphatic alcohols, containing the thietane ring, from dichloro alcohols and alkali metal sulfides [2]. The synthesis of unsaturated thietanes is of interest both in terms of a search for biologically active materials and for the synthesis of novel heterocyclic systems.

Reaction of a mixture (4:1) of trans- and cis-5-methyl-1-chloro-2,3-epoxyhex-5-enes (Ia, b) with an aqueous-alcoholic solution of sodium sulfide gives a 60% yield of the stereoisomers of 2-(2-methylpropen-2-yl)-thietan-3-ols (IIa, b). The structure and configuration of alcohols IIa, b were confirmed by IR and PMR spectroscopic data and by chemical transformations. The IR spectra of IIa, b show intense bands at 3360-3410 cm<sup>-1</sup> which are assigned to OH group absorption and also to bands near 1645 cm<sup>-1</sup> for the C=C bond.

Treatment of the alcohols IIa, b with acetic anhydride in the presence of pyridine gives a 3.3:1 mixture of acetates IIIa and IIIb. Under phase transfer conditions [3] IIa, b and alkyl halides give the ethers IVa, b-VIa, b in around 90% yield. The structures of IIIa, b-VIa, b were also confirmed by IR and PMR spectroscopy. According to the PMR data, ethers VIa, b exist in the ratio 3.7:1. In agreement with the stereochemical composition of ethers III and VI, the alcohols IIa and IIb exist as a 3.5:1 mixture in the starting material.



## **EXPERIMENTAL**

The GLC analysis was carried out on an LKhM-8MD instrument with thermal conductivity detector, steel column (1000  $\times$  3 mm) with 15% PEGA on Dinakhrome, helium bas carrier, and column temperature of 90-140°C. The PMR spectra were recorded on Bruker 250 MHz and Tesla BS-487 (80 MHz) instruments using CCl<sub>4</sub> or CDCl<sub>3</sub> solvent and HMDS internal standard. The IR spectra were taken on Specord-80 or UR-20 instruments as thin films.

Stereoisomers of 2-(2-Methylpropen-2-yl)thietan-3-ol (IIa, b,  $C_7H_{12}OS$ ). A mixture of Ia, b (14.6 g, 110 mmoles) [4],  $Na_2S \cdot 9H_2O$  (31 g, 130 mmoles), water (70 ml), and ethanol (75 ml) was stirred for 4 h at 60-70°C, cooled, extracted with ether (3 × 50 ml), and the extracts dried over MgSO<sub>4</sub>. Removal of ether under reduced pressure gave a residue which was distilled in vacuo to give IIa, b (8.6 g, 60%) with bp 98-99°C (1 hPa),  $n_D^{20}$  1.5228, and  $d_4^{20}$  1.0601. Found MR<sub>D</sub> 41.55; calculated 41.65. IR spectrum: 3446-3360 (OH), 1645 cm<sup>-1</sup> (C=C). PMR spectrum for IIa: 4.70 (1H, s, CH<sub>2</sub>=);

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4.68 (1H, s, CH<sub>2</sub>=); 4.34 (1H, quint, J = 8 Hz, CHO); 3.44 (1H, d, J = 8 Hz, OH); 3.77 (1H, ddd, J = 9, 8, 5 Hz, SCH); 3.08 (2H, d, J = 8 Hz, SCH<sub>2</sub>); 2.56 (1H, dd, J = 15, 5 Hz, CH<sub>2</sub>); 2.24 (1H, dd, J = 15, 9 Hz, CH<sub>2</sub>); 1.66 ppm (3H, s, CH<sub>3</sub>). For IIb: 4.76 (1H, s, CH<sub>2</sub>=), 4.63 (1H, s, CH<sub>2</sub>=); 4.95 (1H, quint, J = 8 Hz, CHO); 3.77 (1H, ddd, J = 9, 8, 5 Hz, SCH); 3.23 (2H, d, J = 8 Hz, SCH<sub>2</sub>); 3.20 (1H, d, J = 8 Hz, OH); 2.61 (1H, dd, J = 15, 5 Hz, CH<sub>2</sub>); 2.44 (1H, dd, J = 15, 9 Hz, CH<sub>2</sub>); 1.68 ppm (3H, s, CH<sub>3</sub>).

3-Acetoxy-2-(2-methylpropen-2-yl)thietane (IIIa, b,  $C_9H_{14}O_2S$ ). A mixture of IIa, b (2 g, 14 mmoles), acetic anhydride (3 g, 29 mmoles), and pyridine (2 g) was heated for 2 h on a steam bath, poured into ice, acidified with HCl, and extracted with chloroform. Removal of solvent gave IIIa, b (2 g, 80%) with bp 102-103°C (7 hPa),  $n_D^{20}$  1.4916, and  $d_4^{20}$  1.0582. Found: MR<sub>D</sub> 51.02; calculated 51.17. PMR spectrum for IIIa: 5.19 (1H, q, J = 8 Hz, CHO); 4.66 (1H, s, H<sub>2</sub>C=); 4.60 (1H, s, H<sub>2</sub>C=); 3.90 (1H, ddd, J = 10, 8, 5 Hz, SCH); 3.13 (2H, d, J = 8 Hz, SCH<sub>2</sub>); 2.54 (1H, dd, J = 15, 5 Hz, CH<sub>2</sub>); 2.26 (1H, dd, J = 15, 10 Hz, CH<sub>2</sub>); 1.93 (3H, s, CH<sub>3</sub>CO); 1.62 (3H, s, CH<sub>3</sub>). For IIIb, 5.69 (1H, q, J = 8 Hz, CHO); 4.72 (1H, s, H<sub>2</sub>C=); 4.58 (1H, s, H<sub>2</sub>C=); 3.80 (1H, q, J = 8 Hz, SCH); 3.46 (1H, t, J = 8 Hz, SCH<sub>2</sub>); 3.20 (1H, t, J = 8 Hz, SCH<sub>2</sub>); 2.45 (2H, d, J = 8 Hz, CH<sub>2</sub>); 1.94 (3H, s, CH<sub>3</sub>CO): 1.62 ppm (3H, s, CH<sub>3</sub>).

**3-Allyloxy-2-(2-methylpropen-2-yl)thietane** (IVa, b,  $C_{10}H_{16}OS$ ). A mixture of IIa, b (2 g, 14 mmoles), allyl bromide (8.2 g, 68 mmoles), benzyltriethylammonium chloride (0.3 g), sodium hydroxide (50%, 7 ml), and benzene (50 ml) is stirred for 5 h at 40-45°C. Appropriate workup gives IVa, b (2.3 g, 90%) with bp 66-67°C (1 hPa),  $n_D^{20}$  1.4976,  $d_4^{20}$  0.9873. Found: MR<sub>D</sub> 54.68, calculated 55.38. IR spectrum: 1645 (C=C), 1024-1292 cm<sup>-1</sup> (COC). PMR spectrum: 5.85 (1H, ddt, J = 17, 10, 6 Hz, HC=); 5.25 (1H, dq, J = 17, 2 Hz, H<sub>2</sub>C=); 5.16 (1H, dq, J = 10, 2 Hz, H<sub>2</sub>C=); 3.90 (2H, m, CH<sub>2</sub>O); for IVa: 4.72 (1H, s, H<sub>2</sub>C=); 4.69 (1H, s, H<sub>2</sub>C=); 4.14 (1H, q, J = 7 Hz, CHO); 3.90 (1H, m, SCH); 3.13 (1H, t, J = 7 Hz, SCH<sub>2</sub>); 3.07 (1H, t, J = 7 Hz, SCH<sub>2</sub>); 2.61 (1H, dd, J = 15, 5 Hz, CH<sub>2</sub>); 2.29 (1H, dd, J = 15, 10 Hz, CH<sub>2</sub>); 1.69 (3H, s, CH<sub>3</sub>); for IVb: 4.78 (1H, s, H<sub>2</sub>C=); 4.64 (1H, s, H<sub>2</sub>C=); 3.90 (1H, m, SCH); 3.76 (1H, q, J = 8 Hz, CHO); 3.38 (2H, t, J = 8 Hz, SCH<sub>2</sub>); 2.61 (1H, dd, J = 15, 5 Hz, CH<sub>2</sub>); 2.29 (1H, dd, J = 15, 10 Hz, CH<sub>3</sub>); 1.71 ppm (3H, s, CH<sub>3</sub>).

**3-Benzyloxy-2-(2-methylpropen-2-yl)thietane** (Va, b,  $C_{14}H_{18}OS$ ). Reaction of alcohol II (9 g, 63 mmoles), benzyl chloride (32 g, 250 mmoles), triethylbenzylammonium chloride (1.3 g), NaOH (50%, 6 g), and benzene (60 ml) similarly to IIIa, b gave Va, b (13.3 g, 91%) with bp 145-146°C (1 hPa),  $n_D^{20}$  1.5476,  $d_4^{20}$  1.0484. Found: MR<sub>D</sub> 70.95, calculated 70.96. According to PMR the Va:Vb isomer ratio was 4:1. PMR spectrum: 7.35 (5H, m, C<sub>6</sub>H<sub>5</sub>); Va: 4.80 (1H, s, H<sub>2</sub>C=); 4.76 (1H, s, H<sub>2</sub>C=); 4.50 (1H, d, J = 12 Hz, OCH<sub>2</sub>); 4.48 (1H, d, J = 12 Hz, OCH<sub>2</sub>); 4.25 (1H, q, J = 8 Hz, CHO); 4.03 (1H, ddd, J = 10, 8, 5 Hz, SCH); 3.22 (1H, t, J = 8 Hz, SCH<sub>2</sub>); 3.07 (1H, t, J = 8 Hz, SCH<sub>2</sub>); 2.62 (1H, dd, J = 14, 5 Hz, CH<sub>2</sub>); 2.33 (1H, dd, J = 14, 10 Hz, CH<sub>2</sub>); 1.74 (3H, s, CH<sub>3</sub>); Vb: 4.87 (1H, s, H<sub>2</sub>C=); 4.76 (1H, s, H<sub>2</sub>C=); 4.47 (2H, s, OCH<sub>2</sub>); 3.83; (1H, q, J = 8 Hz, CHO); 3.50 (2H, t, J = 8 Hz, SCH<sub>2</sub>); 2.68 (2H, d, J = 7 Hz, CH<sub>2</sub>); 1.77 ppm (3H, s, CH<sub>3</sub>).

**2-(2-Methylpropen-2-yl)-3-propargyloxythietane** (VIa, b,  $C_{10}H_{14}OS$ ). Reaction of II (5 g, 35 mmoles), propargyl bromide (8 g, 67 mmoles), NaOH (50%, 7 g), triethylbenzylammonium chloride (1.3 g), and benzene (60 ml) gave VIa, b (5.6 g, 89%) with bp 105-106°C (8 hPa),  $n_D^{20}$  1.5104,  $d_4^{20}$  1.0153. Found: MR<sub>D</sub> 53.73, calculated 53.67. IR spectrum: 3300 and 2100 (C=CH), 1645 (C=C), 1275, 1150-1100 cm<sup>-1</sup> (COC). PMR spectrum: 4.95 (2H, m, H<sub>2</sub>C=); 4.70 (1H, m, CHO); 4.37 (2H, m, CH<sub>2</sub>O); 4.12 (1H, m, SCH); 3.04 (2H, m, SCH<sub>2</sub>); 2.95-2.12 (3H, m, CH<sub>2</sub>, =CH); 1.85 ppm (3H, s, CH<sub>3</sub>).

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